Surface Conditions on Earth Through Time

Nathan D. Sheldon

Earth and Environmental Sciences UNIVERSITY OF MICHIGAN

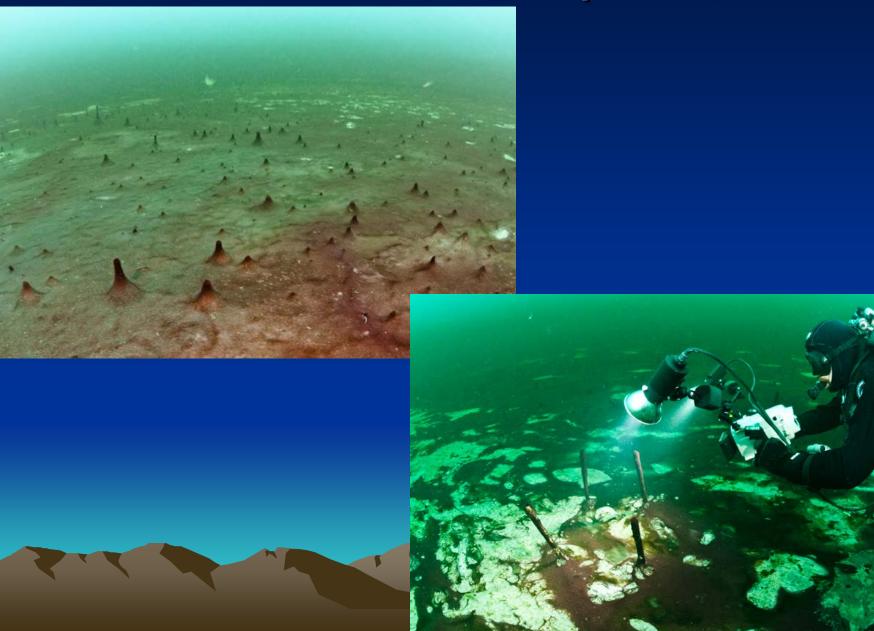
Acknowledgements: Ria Mitchell, Michael Hren



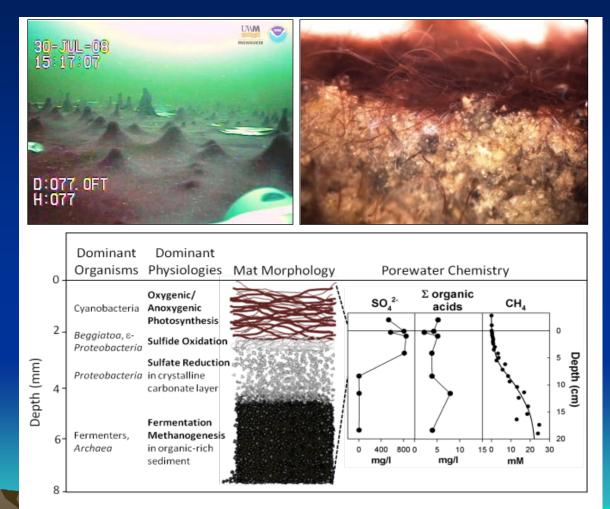
Funding from:

National Science Foundation

Lake Huron near Alpena, MI

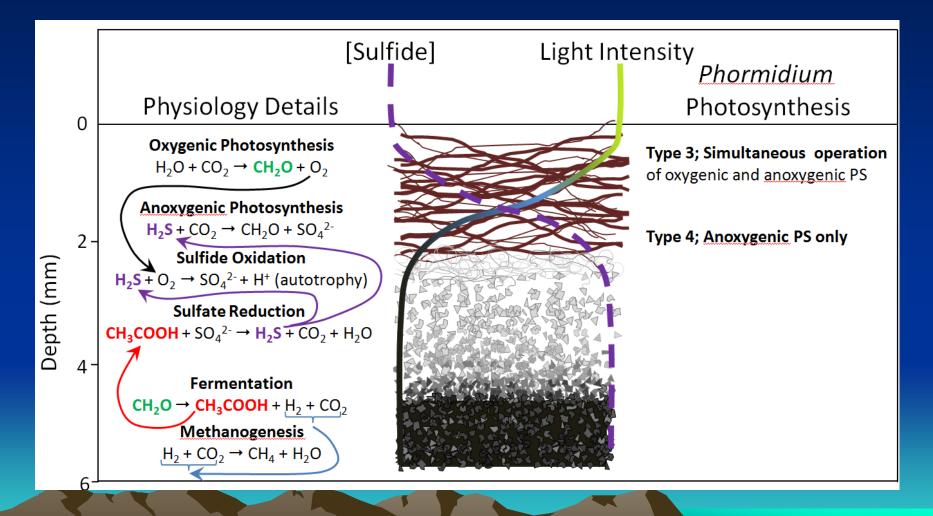


Proterozoic ocean...in modern Lake Huron at 8°C



Voorhies et al. (2012, Geobiology)

Metabolic Flexibility



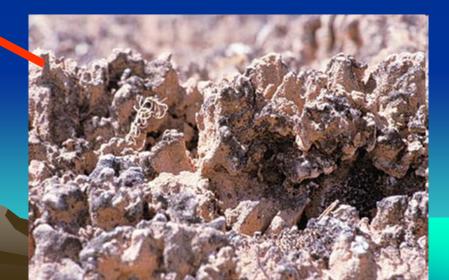
Voorhies et al. (2012, Geobiology)

Life on Earth is Nearly Ubiquitous Soil Crusts and Extremely Dry Environments



Cyanobacterial sheath holding together sand grains (SEM).

- Typically cyanobacteria
- Deserts, Antarctic Dry Valleys

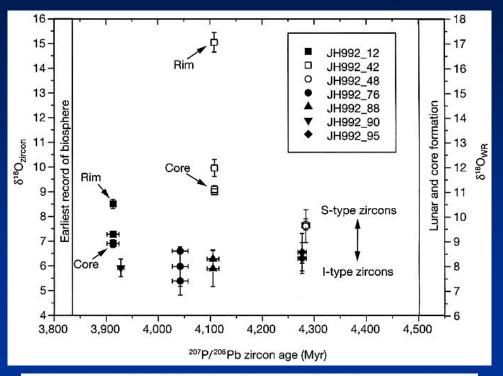


www.soilcrust.org/gallery.htm

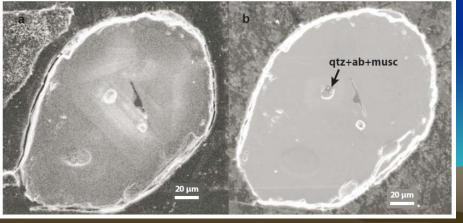
Environmental Limits for Life

- <u>Temperature</u> -10°C 121°C (230 °C?)
- Pressure up to gigapascals (10s of kbars)!
- Salinity up to 5 M NaCl (?)
- <u>pH</u> ~0-12 (?)
- Liquid water available
- <u>O₂ available</u>
- Only bacteria and archaea can withstand the highest T, P, etc.

Evidence for Liquid Water from Zircons



 Zircons from Jack Hills (Australia) and other Hadean sites record δ¹⁸O values that indicate weathering in liquid water back to 4.4 Ga







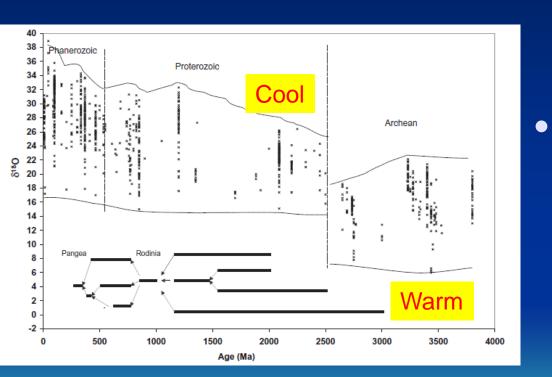
We simultaneously know much less about the Earth's surface and much more about the history of its surface as compared with Mars!

Liquid water is nearly unique to Earth, but is it sufficient in and of itself to result in a habitable planet?

Habitability on the Earth

- To a first order, it depends on just three things:
 - Surface temperature (i.e., climate) was probably never substantially different than the present-day Earth
 - pO₂ (i.e., redox potential; constrains other greenhouse gases) rose twice, both times fundamentally changing the biosphere
 - pCO₂ (i.e., the organic-inorganic C cycle) is the ultimate control of Earth surface conditions and has both shaped the biosphere and been shaped by the biosphere

I. Arguments for Very Warm Early Earth

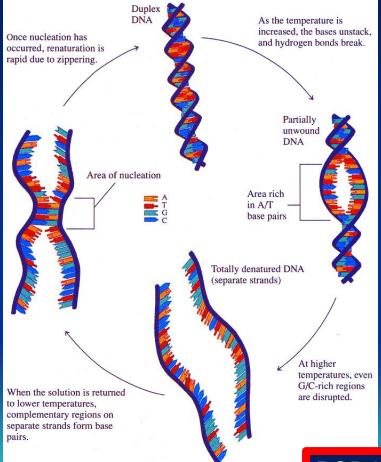


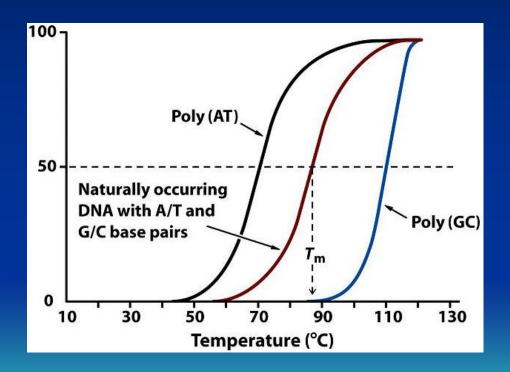
Knauth (2005, P-3)

Chert is an early diagenetic product derived from amorphous silica (cf. quartz)

Archean (pre 2.5 Ga) cherts have very different $\delta^{18}O$ than Proterozoic and Phanerozoic cherts If sea water chemistry hasn't changed, this implies very warm early oceans (55-85°C), which implies high CO₂, salinity, and P_{atm} s this reasonable?

Biochemical Constraints on Ocean Temperatures

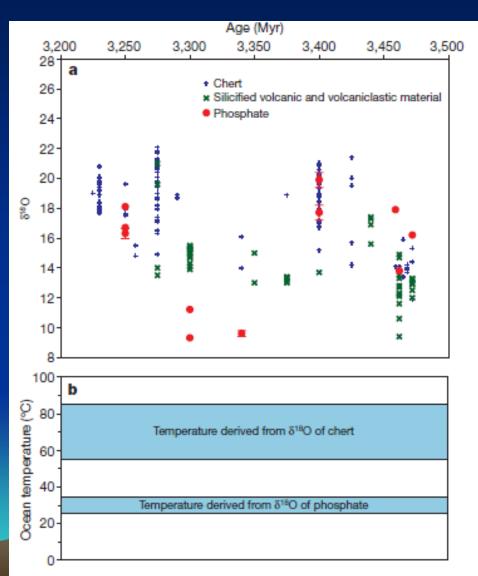




PCR is typically performed at 94°C, but depending on the base pair make up, the melting temperature of DNA/RNA can be as low as 70°C.

Images from Sandwalk

What do other minerals indicate?

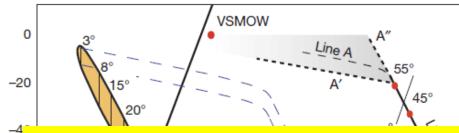


Blake et al. (2011, Nature) used the same rocks as Knauth and looked instead at phosphates

 Instead of warm temperatures, they found temperatures similar to the modern ocean

 Gypsum gives a similar constraint

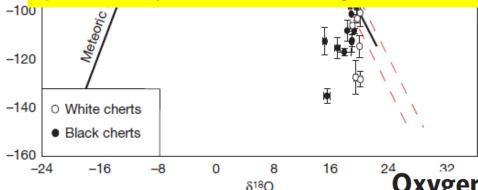
Chert Revisited: Adding δD Lowers Temperatures



ß

 Including δD from cherts adds a further constraint and precludes a straight

Thus, the weight of both biological and mineral alues evidence is consistent with temperate conditions (<35°C) for virtually all of Earth's history



Cherts have been allered

• $T_{max} < 40^{\circ}C$

Oxygen and hydrogen isotope evidence for a temperate climate 3.42 billion years ago

M. T. Hren¹[†], M. M. Tice² & C. P. Chamberlain³

Vol 462 12 November 2009 doi:10.1038/nature08518

II. History of pO₂

- Two competing models have developed:
 - Dimroth-Ohmoto model suggests that atmospheric pO₂ has not changed substantially during Earth's history
 - Fe mobility during surface weathering hasn't changed dramatically
 - BIFs (and other ore deposits) could all be hydrothermal, and therefore, tell us very little about surface conditions
 - C cycle changes don't correspond to proposed pO₂ changes and the two cycles should be linked

Additional Feedbacks in the Long-term Carbon Cycle

For organic matter:

$$CO_2 + H_2O \xrightarrow[weathering]{\text{burial}} CH_2O + O_2$$

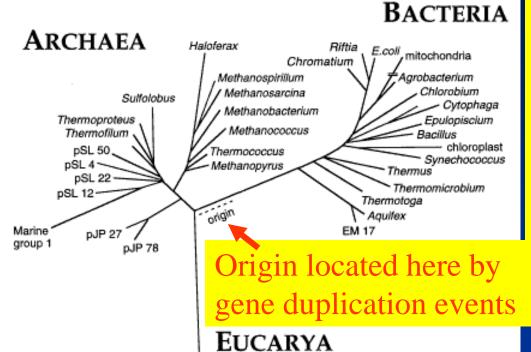
For pyrite:

 $2Fe_{2}O_{3} + 16Ca^{2+} + 16HCO_{3}^{-} + 8SO_{4}^{2-}$

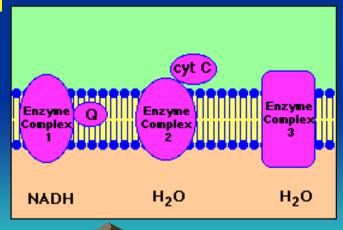
burial weathering $4FeS_2 + 16CaCO_3 + 8H_2O + 15O_2$

Berner and Canfield (1989, Am. J. Sci.)

Universal Tree of Life (based on RNA)



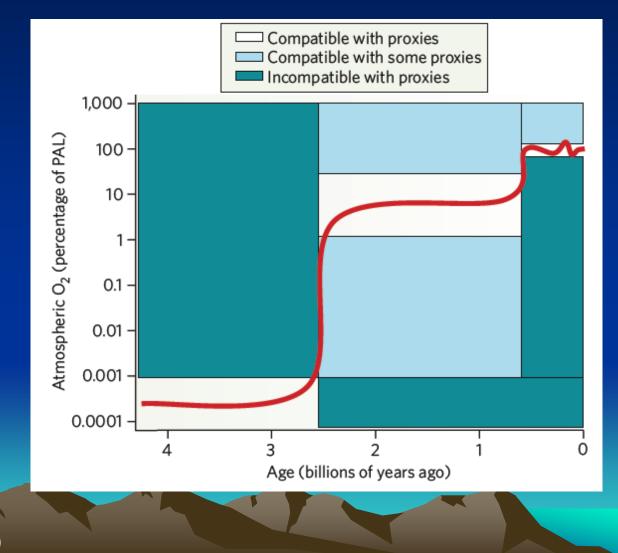
The last common ancestor lived on a planet with redox gradients extending down to molecular O_2 ! Genomics: the last common ancestor had at least 4 respiratory Chains terminating in: O_2 , nitrate, sulfate, and S. (Castresana & Moreira, 1999)



II. History of pO₂

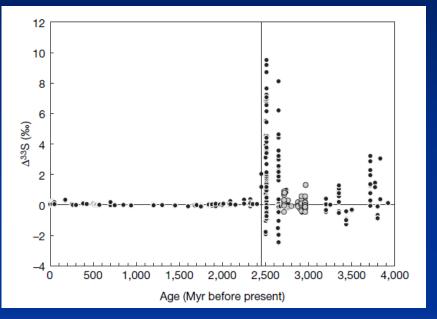
- Two competing models have developed:
 - Dimroth-Ohmoto model suggests that atmospheric pO₂ has not changed substantially during Earth's history
 - Cloud-Walker-Holland model suggests that early Earth was essentially anoxic and that pO₂ rose substantially early on in the Proterozoic → model is now canonical

Proxy Constraints on pO₂



Kump (2008)

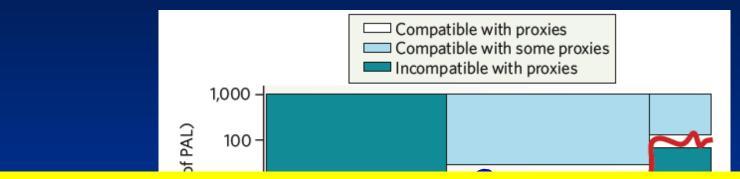
"Great Oxidation Event"



Farquhar et al. (2007, Nature)

- Various lines of evidence (red beds, detrital uraninite, paleosols, etc.) indicate a singular increase in pO₂
- MIF-S disappears after 2.45 Ga
- This implies a rise in pO₂ to > 10⁻⁵ bars
- There may have been a "whiff" of oxygen prior to the GOE
- Consequences: loss of CH₄, increase in biological niches

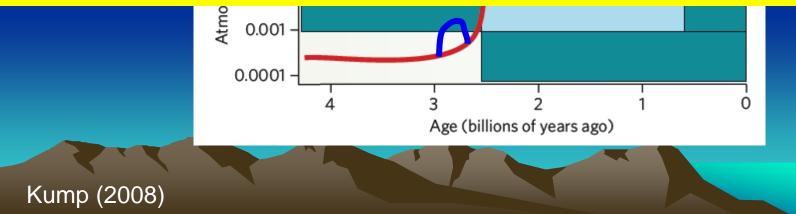
Proxy Constraints on pO₂



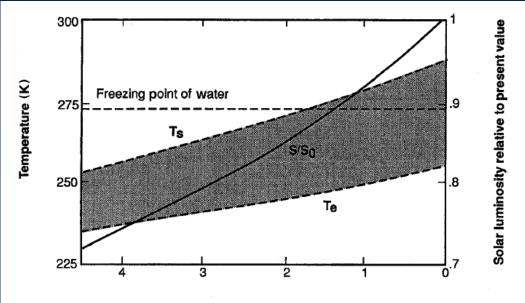
Open questions include:

- 1) What triggered the GOE?
- 2) What happened to pO_2 and ocean chemistry during the "boring billion?"

3) What triggered the second rise in pO_2 that ultimately gave rise to animals?



III. "Faint Young Sun" Paradox



Billions of years before present

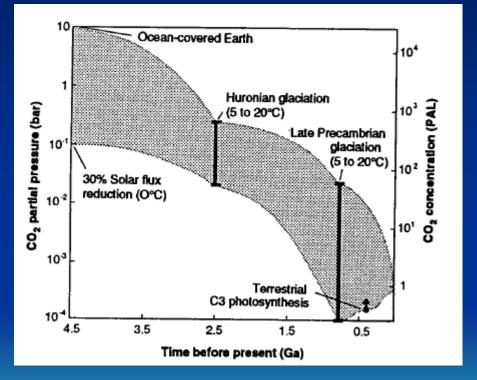
 The Earth's surface appears to have been temperate

 The Sun was likely only 70% as luminous early in its history

 Therefore, significantly higher greenhouse gas levels are required

Kasting (1988, Scientific America)

pCO₂-only Solution to FYS



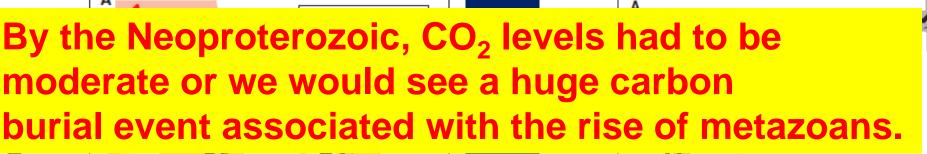
Kasting (1993, Science)

 Requires at least 0.1 bars of CO₂, whereas the pre-Industrial atmosphere had 10^{-3.5} bars

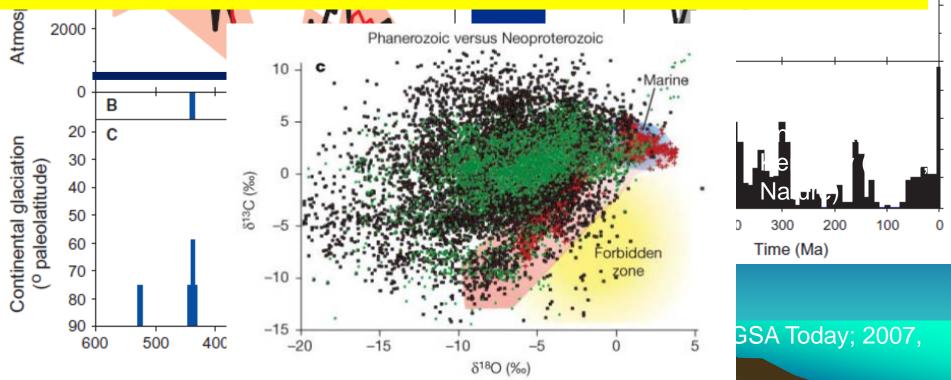
 This would result in extremely acidic rain and should have left an isotopic signature of the burial of that

carbon

Constraints from the last 500 Ma



8000



Part IV. Constraining CO₂ History
A. Equilibrium models for CO₂ and their limitations

B. A mass balance model for pCO₂ based on paleosols

C. Towards a Precambrian CO₂ Curve

D. 1-D Climate model results

A. Equilibrium Model: Rye et al. (1995, Nature)

- Greenalite is a good analogue mineral for the actual assemblage
- In situ measurements of CO₂ from BIFs give comparable values to the calculations
- If CO₂ levels were high enough, siderite would form, thus no siderite in Precambrian paleosols implies lower CO₂ levels

 $\begin{array}{l} \mathsf{Fe}_3\mathsf{Si}_2\mathsf{O}_5(\mathsf{OH})_4 + 2\mathsf{H}_2\mathsf{O} + 3\mathsf{CO}_2 = 3\mathsf{FeCO}_3 + 2\mathsf{H}_4\mathsf{SiO}_4\\ \text{greenalite} & \text{siderite} \end{array}$

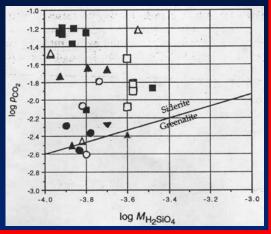
$$P_{CO_{2}} = \sqrt[3]{\frac{a_{H_{4}SiO_{4}}^{2}}{K_{RXN}}}$$

Atmospheric carbon dioxide concentrations before 2.2 billion years ago

Rob Rye, Phillip H. Kuo & Heinrich D. Holland

Department of Earth and Planetary Sciences, Harvard University, 20 Oxford Street, Cambridge, Massachusetts 02138, USA

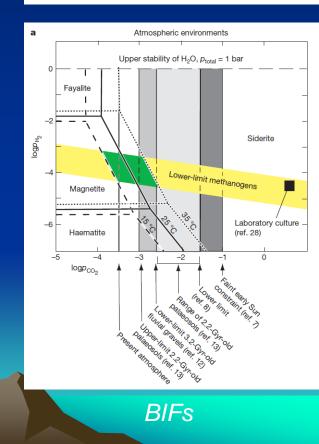
Paleosols



Equilibrium Models for Precambrian pCO₂

No climate paradox under the faint early Sun

Minik T. Rosing^{1,2,4}, Dennis K. Bird^{1,4}, Norman H. Sleep⁵ & Christian J. Bjerrum^{1,3}



A lower limit for atmospheric carbon dioxide levels 3.2 billion years ago

Angela M. Hessler*, Donald R. Lowe, Robert L. Jones & Dennis K. Bird

Department of Geological and Environmental Sciences, Stanford University, Stanford, California 94305-2115, USA

Carbonate Rinds

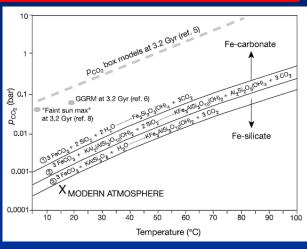
Evidence from massive siderite beds for a CO₂-rich atmosphere before \sim 1.8 billion years ago

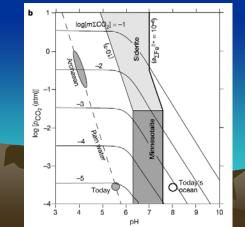
Hiroshi Ohmoto $^{\rm 1}$, Yumiko Watanabe $^{\rm 1}$ & Kazumasa Kumazawa $^{\rm 2}$

¹Astrobiology Research Center of the NASA Astrobiology Institute and Department of Geosciences, The Pennsylvania State University, University Park, PA 16802, USA

²Oyo Corporation, Miyazaki Branch, Oshima-cho, Miyazaki City, 0995-61, Japan

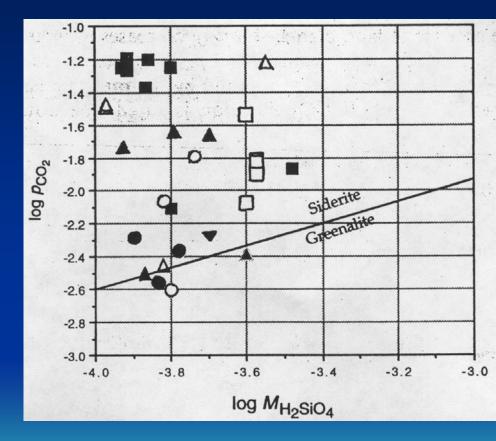






Problems

- BIF "proof" is irrelevant
- Thermodynamic data used is out of date
- First occurrence of siderite in a Precambrian paleosol is <1Ga ago
- Greenalite doesn't form authigenically in soils and isn't present at all in the Precambrian paleosols they examined



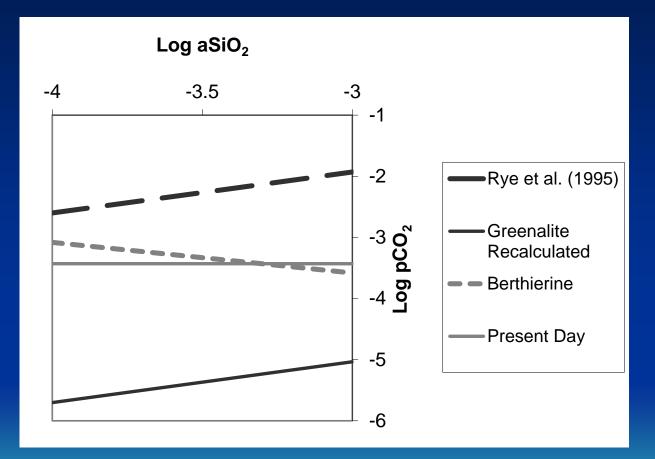
Rye et al. (1995, Nature)

Alternative Equilibrium Model: Berthierine

• Berthierine is a part of the mineral assemblage of the Hekpoort paleosol (Retallack, 1986; Retallack and Krinsley, 1993) and forms under reducing conditions (Sheldon and Retallack, 2002)

$$P_{CO_2} = \sqrt{\frac{K_{RXN}}{a_{SiO_{2(aq)}}}}$$

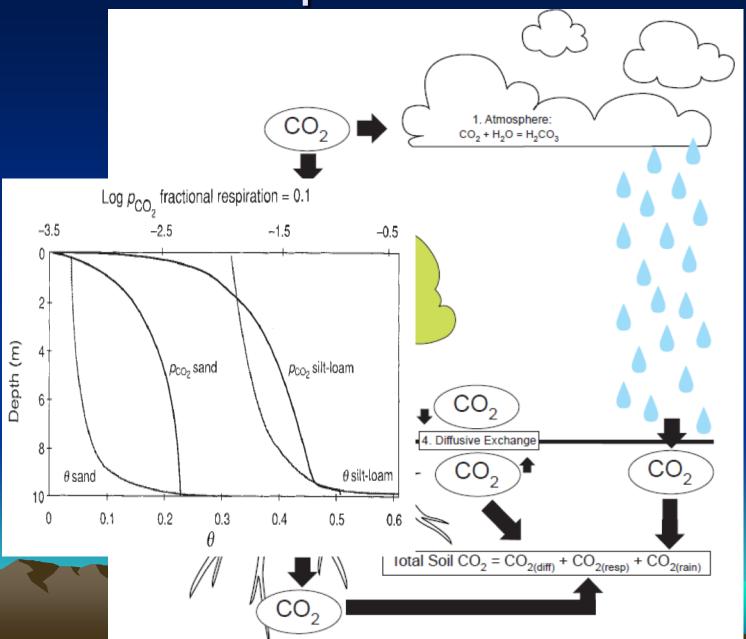
Recalculated pCO₂



Sheldon (2006, Precambrian Research)

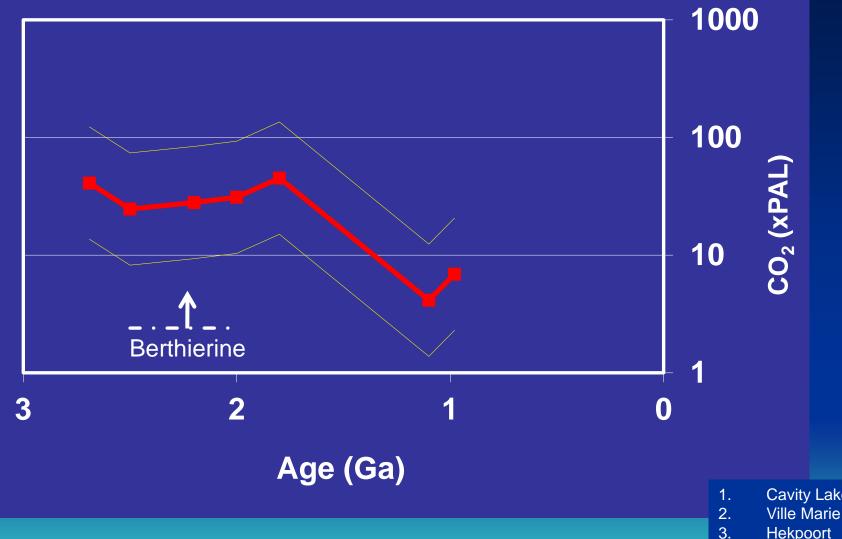
If you take nothing else away from this part of the talk, please realize that thermodynamic models only provide **very, very weak guidance** about pCO₂ levels!

Conceptual Framework



Keller and Wood (1993, Nature)

C. Towards a Precambrian CO₂ Curve



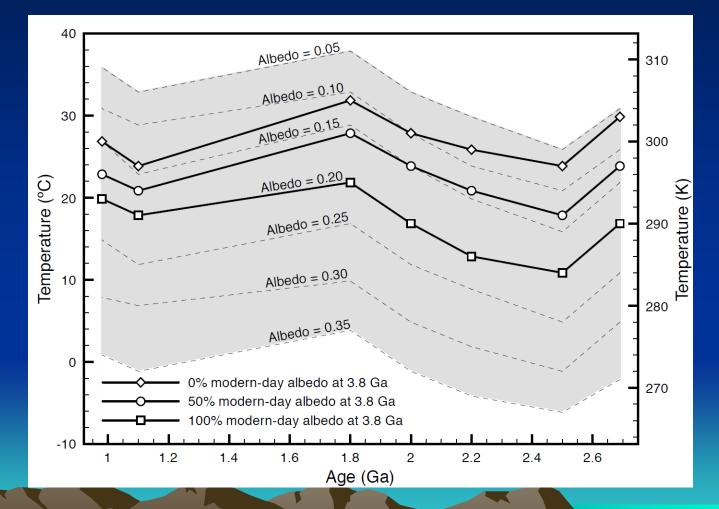
Compiled from Sheldon (2006, PC Res.), Mitchell and Sheldon (2010, PC Res.), and Driese et al. (2011, PC Res.)

- **Cavity Lake**
- Hekpoort
- 4. Drakenstein 5.
 - Flin Flon
- Sturgeon Falls 6. 7.
 - Sheigra

D. 1-D Climate Models (w/ S. Domagal-Goldman)

- Use revised upper limit pCO₂ curve from paleosols
- Use two coupled models, one for photochemistry (CO₂, CH₄, O₂, O₃, C₂H₆, H₂O, particles (mostly S compounds), which is then fed into a 1-D climate model to calculate temperatures
 - Sim. to Haqq-Misra et al. (2008, Astrobiology) and Domagal-Goldman et al. (2011, Astrobiology)
- Assumptions:
 - Well-mixed atmosphere
 - After 2.4 Ga ago, 5% O₂ concentration and modern CH₄ flux to atmosphere; no O₂ flux to atmosphere, instead controlled by other species
 - Modern-day volcanic sources of H_2 , SO_2 , and H_2S
 - Variable albedo and different continental growth scenarios considered

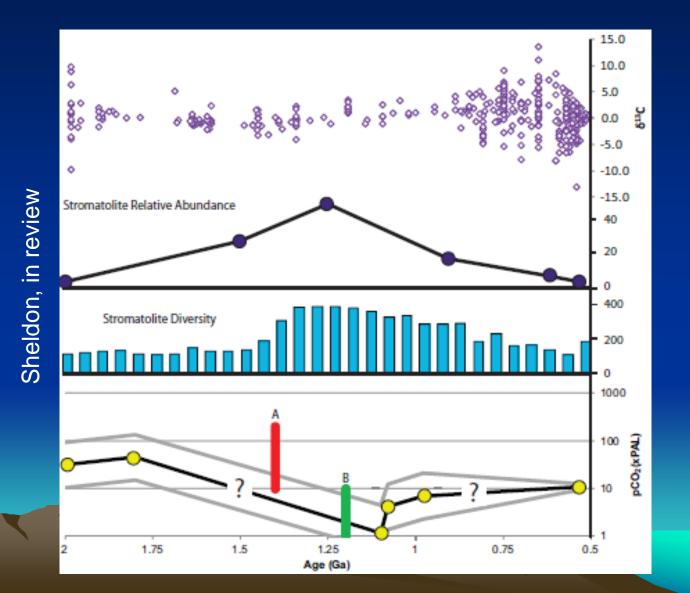
1-D Climate Model Results



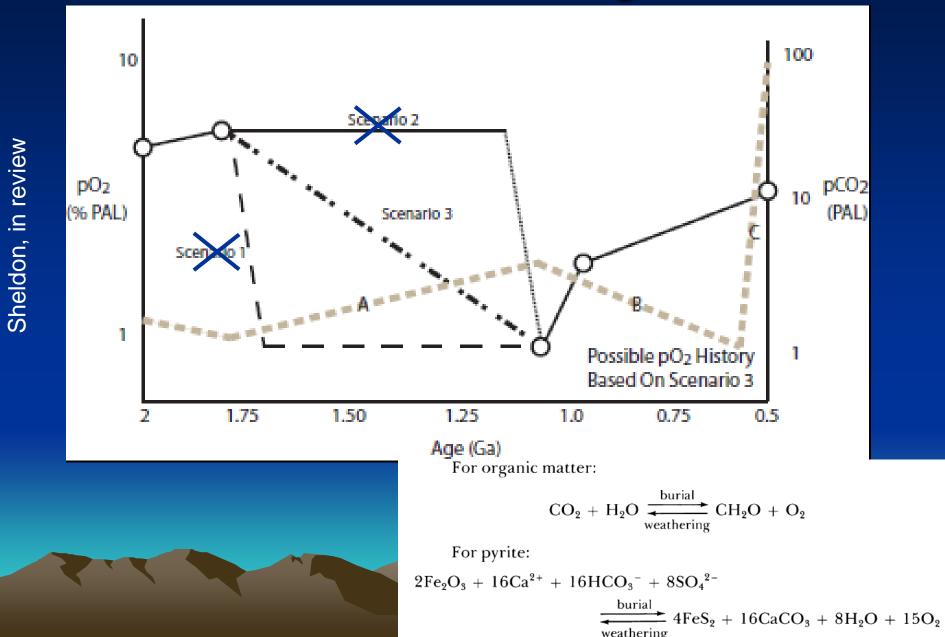
Domagal-Goldman and Sheldon (in prep)

So, if moderate pCO₂ were able to maintain equable conditions, it really must have been a "boring billion"...or was it?

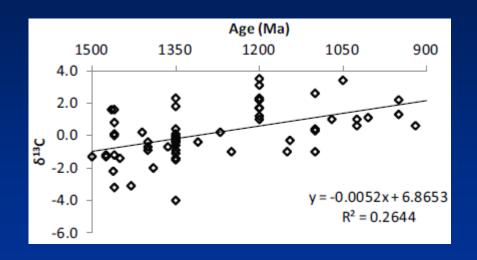
Earth's Middle Ages



Earth's Middle Ages II



Earth's Middle Ages III



$$F_{org} = F_{tot} \frac{\delta^{13} C_{carb} - \delta^{13} C_{input}}{\Delta^{13} C}$$

 If scenario three is correct, then we should see evidence for gradual C burial

 Based upon simple mass balance, 6x10¹⁴ moles of excess C are buried

 Corresponds to rise of calcifying algae, which store C in their

sheaths

Conclusions

- Temperatures have been equable since at least 3.5 Ga ago, but probably 4.0 Ga
- Oxygen levels rose twice, between 2.45-2.22 Ga and 0.7-0.54 Ga
- Another greenhouse gas such as methane had to be present to account for the "faint young Sun" paradox, but both glaciation and equable conditions are possible with known constraints; albedo could be moderately important
- The "boring billion" actually represents a time of significant environmental change